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homogeneous mixture of two or more components in which the particle size is smaller than 1 nm
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solution all the components appear as a single phase there is particle homogeneity i e particles are
evenly distributed**

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process of forming a solution 3 chemistry the state of being dissolved esp in the phrase in solution
4 chemistry a mixture of two or more substances in which one or more components are present as
small particles with colloidal dimension colloid a colloidal solution**

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a problem b an answer to a problem explanation specifically a set of values of the variables that
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used or done to deal with and end a problem something that solves a problem medication may not
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particles may be visible in a suspension a solution does not cause beams of light to scatter**

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mixture of two or more substances in relative amounts that can be varied continuously up to what
is called the limit of solubility the term solution is commonly applied to the liquid state of matter
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homogeneous mixture of two or more substances a solution may exist in any phase a solution
consists of a solute and a solvent the solute is the substance that is dissolved in the solvent the
amount of solute that can be dissolved in solvent is called its solubility**

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solving an explanation or answer the solution is as good as any other mathematics the process of
determining the answer to a problem**

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countable solve deal with a problem a way of solving a problem or dealing with a difficult situation
solve the best solution would be for them to separate solution to for there are no simple solutions**

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the solution if you dissolve a solid into a liquid you ve created a different kind of solution**

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